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<p>(54) Title: METHOD OF OBTAINING (22R) DIASTEREoisomer OF BUDESONIDE</p> <p>(57) Abstract</p> <p>By the method according to the invention condensation of 11β,16α,17α,21-tetrahydroxy-1,4-pregnadiene-3,20-dione 21-acetate with n-butyric aldehyde is carried out, in the known way, in the medium of hydrofluoric acid of concentration of 70-80 %. The isolated crude condensation product is crystallized from ethanol and obtained 21-acetate of budesonide (22R) of at least 95 % content is hydrolyzed, and the product thus obtained is crystallized from ethyl acetate.</p> <p style="text-align: center;">10</p>		

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⁺ Any designation of "SU" has effect in the Russian Federation. It is not yet known whether any such designation has effect in other States of the former Soviet Union.

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Method of obtaining (22R) diastereoisomer of budesonide

The subject of the invention is a method of obtaining diastereoisomer of (22R) budesonide. Budesonide, i.e. (22R,S)-16 α ,17 α -butylidenedioxy-11 β ,21-dihydroxy-1,4-pregnadiene-3,20-dione is a mixture of diastereoisomers (22R) and (22S) differing in the position of an acetal chain (fig.1 and 2). Both compounds are active glucocorticoids applied in a mixture (1:1) in pharmaceutical forms: antiasthmatic aerosol or antiallergic ointment. (22R) diastereoisomer is 2-3 times more active pharmacologically than (22S) diastereoisomer [Brattsand R.; Eur.J.Res.Dis.63, suppl. 122,62-73 (1982)].

15 In pharmacotherapy a tendency is presently observed to apply optically active compounds and not racemic mixtures. In the case of budesonide both isomeric forms are pharmacologically active but their metabolism is different [Andersson P. et al.; Xenobiotica 17,35-44 (1987)]. Testing of action of a drug being a pure chemical individual is considerably easier than that of a mixture.

A known method of obtaining budesonide, that is a mixture of diastereoisomers (22R) and (22S) consists in condensation of 11 β ,16 α ,17 α ,21-tetrahydroxy-1,4-pregnadien-3,20-dione with n-butyric aldehyde in the presence of strong inorganic acids, e.g. perchloric acid, in organic solvents [FRG patent 2323216(1973)].

Those skilled in the art know also a method of obtaining a mixture of diastereoisomers (22R) and (22S), in which isomer (22R) is in majority, even up to 90%. The said method consists in condensation, discussed above, conducted in the presence of hydrofluoric acid of concentration of 48-70% or concentrated hydrochloric acid [Eur.pat.appln.164636 (28.05.85)].

According to the known method separation of a mixture of diastereoisomers (22R) and (22S) is carried out on a

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column packed with Sephadex LH20 [Thalen A., Nylander B.:
19,247-266(1982)]. The said method requires the use of a
very big amount of solvents and long separation time, and
for these reasons it is technologically inconvenient.
5 expensive and time-consuming.

Unexpectedly it has appeared out that the product
obtained after condensation in the form of 21-acetate
containing at least 80% of diastereoisomer (22R), after
crystallization from ethanol, and then hydrolyzed and
10 crystallized from ethyl acetate yields pure
diastereoisomer (22R) with an admixture of
diastereoisomer (22S) of 1% at the very most. The content
of diastereoisomers is determined by the HPLC method.
From post-crystallization filtrates containing a mixture
15 of diastereoisomers of a composition of 8:2 or 7:3 one
may separate budesonide 21-acetate, crystallize it again
and after hydrolysis obtain pure diastereoisomer (22R).

By the method according to the invention condensation
of 11 β ,16 α ,17 α ,21-tetrahydroxy-1,4-pregnadiene-3,20-dione
20 21-acetate with n-butyric aldehyde is carried out, in the
known way, in the medium of hydrofluoric acid of
concentration of 70-80%. The isolated crude condensation
product is crystallized from ethanol and obtained
21-acetate of budesonide (22R) of at least 95% content is
25 hydrolyzed, and the product thus obtained is crystallized
from ethyl acetate to obtain (22R) diastereoisomer of
budesonide of at least 99% content.

An advantageous effect of the invention is that (22R)
diastereoisomer of budesonide having the content of at
30 least 99% is obtained at a yield of about 60%. The
product obtained by the method according to the invention
is an active substance of antiasthmatic aerosol and its
content in a dose of 158 μ g is equivalent to 200 μ g of
budesonide of a mixture (1:1) of diastereoisomers (22R
35 and 22S). The method according to the invention is
characterized by simplicity of procedure, enables saving
organic solvents and is less labour-consuming.

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The after-mentioned example illustrates the invention without limiting its scope.

Example.

To a mixture of 7 ml of 75% hydrofluoric acid and 5 0.52 ml of n-butyric aldehyde, cooled to 0°C, 3.5 g of 11 β ,16 α ,17 α ,21-tetrahydroxy-1,4-pregnadiene-3,20-dione 21-acetate is added in and stirred for 3 hours at maintenance of the temperature of 0°C. The obtained dark-red solution is poured into 50 ml of water with ice, 10 neutralized by concentrated ammonia and extracted by chloroform. From the extract chloroform is distilled off under diminished pressure, and the residue is crystallized from ethanol. 2.2 g of 21-acetate of (22R) diastereoisomer of budesonide of $[\alpha]_D^{22} +106^\circ$ (c=1, CH₂Cl₂) 15 is obtained, the content of 21-acetate of (22S) diastereoisomer of budesonide is 5%. The product is suspended in 40 ml of methanol, cooled to 0°C, 2.5 ml of 10% aqueous solution of potassium carbonate is added thereto, and is stirred under nitrogen for 1 and 1/2 20 hours, at maintenance of the temperature of 0°C, after this it is neutralized with acetic acid, methanol is evaporated and (22R) diastereoisomer of budesonide is isolated either by filtering off or by extraction with chloroform. The separated product is crystallized from 25 ethyl acetate and 1.7 g of (22R) diastereoisomer of budesonide is obtained, of melting point of 245-250°C (decomp.) $[\alpha]_D^{22} +117.5^\circ$ (c=1, CH₂Cl₂); $a_{1\text{cm}}^{1\%}$ 350 at 242 nm. The content of (22S) diastereoisomer of budesonide determined by the the HPLC method is 1%.

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Patent claim

Method of obtaining of (22R) diastereoisomer of budesonide, consisting in a condensation reaction of
5 11 β ,16 α ,17 α ,21-tetra-hydroxy-1,4-pregnadiene-3,20-dione
21-acetate with n-butyric aldehyde conducted in the
medium of hydrofluoric acid of concentration of 70-80%,
characterized in that the isolated crude product of
condensation is crystallized from ethanol and obtained
10 21-acetate of budesonide (22R) of at least 95% content is
hydrolyzed, and the product thus obtained is crystallized
from ethyl acetate.

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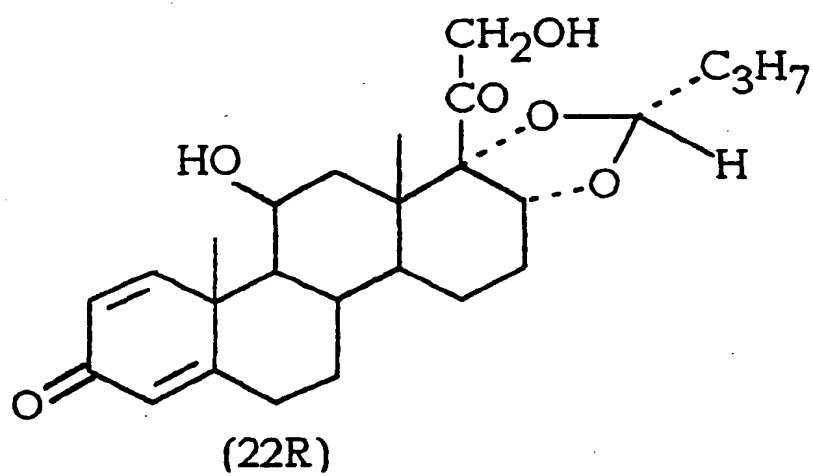


Fig. 1

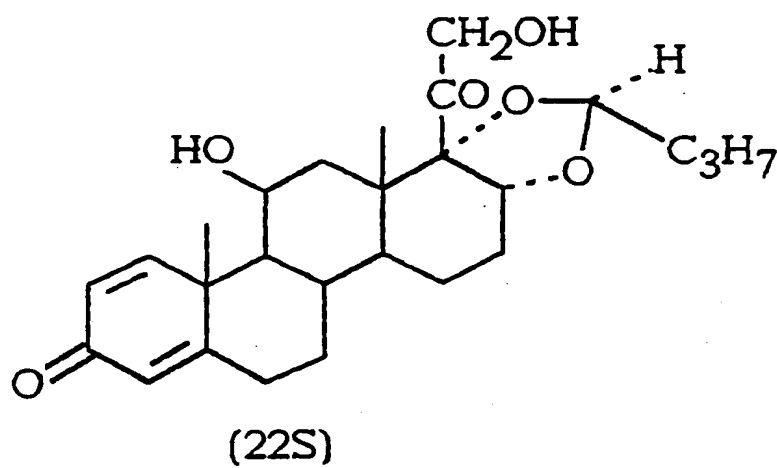


Fig. 2

INTERNATIONAL SEARCH REPORT

International Application No

PCT/PL 91/00016

I. CLASSIFICATION OF SUBJECT MATTER (If several classification symbols apply, indicate all) ⁶ According to International Patent Classification (IPC) or to both National Classification and IPC Int.Cl. 5 C07J71/00				
II. FIELDS SEARCHED <div style="display: flex; justify-content: space-between;"> Minimum Documentation Searched⁷ Classification Symbols </div> <table style="width: 100%; border-collapse: collapse;"> <tr> <td style="width: 30%; border: 1px solid black; padding: 5px;">Int.Cl. 5</td> <td style="border: 1px solid black; padding: 5px;">C07J</td> </tr> </table>			Int.Cl. 5	C07J
Int.Cl. 5	C07J			
Documentation Searched other than Minimum Documentation to the Extent that such Documents are Included in the Fields Searched ⁸				
III. DOCUMENTS CONSIDERED TO BE RELEVANT⁹				
Category ¹⁰	Citation of Document, ¹¹ with indication, where appropriate, of the relevant passages ¹²	Relevant to Claim No. ¹³		
A	EP,A,0 164 636 (SICOR SOCIETA ITALIANA CORTICOSTEROIDI S.P.A.) 18 December 1985 cited in the application see page 8; example 1 ---	1		
A	US,A,3 996 359 (AB BOFORS) 7 December 1976 see the whole document ---	1		
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IV. CERTIFICATION				
Date of the Actual Completion of the International Search <div style="text-align: center; font-weight: bold;">16 MARCH 1992</div>	Date of Mailing of this International Search Report <div style="text-align: center; font-weight: bold;">24. 03. 92</div>			
International Searching Authority <div style="text-align: center; font-weight: bold;">EUROPEAN PATENT OFFICE</div>	Signature of Authorized Officer <div style="text-align: center;">WATCHORN P.W. <i>Peter Watchorn</i></div>			

**ANNEX TO THE INTERNATIONAL SEARCH REPORT
ON INTERNATIONAL PATENT APPLICATION NO. PL 9100016
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		JP-B- 2013680	04-04-90
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